Lab 59: Water as a Solvent

Key Words: Salinity, Solvent

Getting Started:





1. Both of the containers above contain water, but only one of them would quench your thirst. What do you think is different about them?

2. Read the introduction and Challenge to Activity 59, "Water as a Solvent," in your Student Book.

Procedure:

- 1. Read Procedure Steps 1-8 in your Student Book.
- 2. Watch the LABsent video (found here: https://vimeo.com/152305878). Record your observations in the table provided. The video will mix the following liquids and solids.

Mixing Liquids and Solids				
Cup	Liquid	Soltd		
1	15 drops of water	2 level scoops of sodium chloride		
2	15 drops of ethanol	2 level scoops of sodium chloride		
3	15 drops of mineral oil	2 level scoops of sodium chloride		
4	None	2 level scoops of sodium chloride		
5	15 drops of water	2 level scoops of calcium chloride		
6	15 drops of ethanol	2 level scoops of calcium chloride		
7	15 drops of mineral oil	2 level scoops of calcium chloride		
8	None	2 level scoops of calcium chloride		

Data Table for Observations

Cup No.	Liquid Contents	Observations of Liquid	Observations of Liquid and Solid Mixed Together	Amount of Solid Dissolved in Liquid
1	water			
2	ethanol			
3	mineral oil			
4	no liquid			
5	water			
6	ethanol			
7	mineral oil			
8	no liquid			

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Follow-Up: 1. Can you tell how well a solid will dis	ssolve in a solvent based on	what the solvent looks like?
2. Look at Student Sheet 50.2, "Scientif career that you chose to focus on an		
Analysis Questions:		
1. A liquid that has a solid dissolved in a. What are the solvents?b. In which solvent did the solid	•	vent). In this investigation:
2. All water on the earth contains some salt, with sodium chloride (table salt also found on earth's surface. Would	t) being the most common d	issolved salt. Calcium chloride is
3. Water is sometimes called the unive Support your answer with evidence		ou think this statement means.

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4. Do you think most of the wa	ter on the earth is sal	ltwater or fresh	water? Why?	
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